Problems – Set 1: Acids and bases

Problem 1

Put in order of acidity the following compounds and justify: CH_4 , H_2O , NH_3 , H_2S , HBr and HF.

Problem 2

Give the order of basicity:

- a) pyridine, ammoniac, triethylamine, aniline
- b) 4-aminopyridine, 2-6-dimethylpyridine, 2-chloropyridine
- c) NaOEt, NaOtBu, NaOH, NaOnBu, NaOMe

Problem 3

Place the following series of compounds in order of acidity and provide the explanation.

- (a) ClCH₂CO₂H, FCH₂CO₂H, BrCH₂CO₂H, ICH₂CO₂H;
- (b) CH₂CHCO₂H, HCCCO₂H, CH₃CH₂CO₂H;
- (c) CH_3CO_2H , $CH_3CH_2CO_2H$, $(CH_3)_3CCO_2H$;
- (d) CH₃CO₂H, FCH₂CO₂H, F₂CHCO₂H, F₃CCO₂H.

Problem 4

How many resonance structures exist for the anion of *ortho*-cresol, 2-nitrophenol, and 3-nitrophenol? Classify the order of acidity of these compounds.

Problem 5

Explain the following observations:

